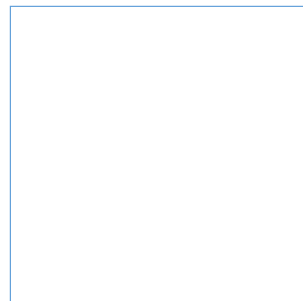
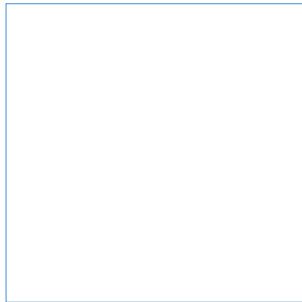
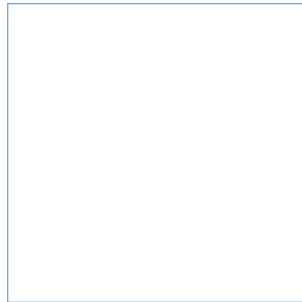
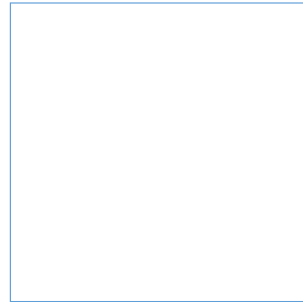


# Scharlau



## Standard buffer solutions for pH-meter calibration

Reliable ■ Traceable to NIST ■ Tested according to DIN 19266 and DIN 19268 ■ Supplied with certificate

*pH value is probably the most common of all routinely performed measurements in laboratories. Since pH-value affects all chemical and biochemical reactions, it is very important to have a reliable measurement.*

*pH-meters measure the voltage developed between two electrodes immersed in the sample and compare that value to a calibration derived from the same electrode pair and known standards. These standard buffer solutions must be accurate and reliable.*

*Scharlau standard buffer solutions are precise, reliable and directly traceable to NIST. They are measured performing a five-point calibration according to DIN 19268. Calibration standards are prepared according to DIN 19266.*



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<p>pH at 20 °C.....4,00 uncertainty ± 0,01 composition per litre to 10,21g potassium hydrogen phthalate Content: 500ml</p> <p>Standard buffer solutions are prepared with gravimetric and volumetric procedures. The batch value is determined by measurement with a combination glass electrode against 15 standard calibration solutions according to DIN 19266.</p> <p>pH..... 20 °C..... 25 °C..... 30 °C..... 35 °C..... 40 °C..... 45 °C..... 50 °C..... 55 °C..... 60 °C..... 65 °C..... 70 °C..... 75 °C..... 80 °C..... 85 °C..... 90 °C..... 95 °C..... (See other parameters in our general catalog)</p> <p>Date of opening</p> <input type="text"/>	<p><b>SO1004</b> <b>Buffer solution pH = 4,00 (20 °C) (Potassium hydrogen phthalate)</b> Solución tampón pH = 4,00 (20 °C) (Potasio hidrógenoftalato) Pufferlösung pH = 4,00 (20 °C) (Kaliumhydrogenphthalat) Solution tampon pH = 4,00 (20 °C) (Potassium hydroginofthalate) Soluzione tampone pH = 4,00 (20 °C) (Potassio italato acido)</p> <p>D=1,01 g/cm<sup>3</sup></p>	<p><b>Traceability</b> This pH buffer solution is traceable to Standard Reference Material from NIST SRM 185a Potassium hydrogen phthalate, SRM 186a Potassium Buffer, SRM 188 Potassium hydrogen tartrate, SRM 189a Potassium tartrate and SRM 213a Calcium carbonate</p> <p><b>Storage and use</b> For pH-meter calibration. Once the bottle is opened, store tightly closed at room temperature. Avoid exposure to light. Remember to write the date of opening in the space provided in the label for 90 days. Never introduce the electrode in the bottle for measurements. Never pour the used solution back in the bottle.</p> <p>Scharlau Chemie S.A. Calle Pineda, 21 - P. 1 - 46100 Sagunto (Valencia) 08181 Santmenni (Spain) Tel: 0034 91 8111111 Made in the European Union. CE Label. 15 months CE. CE Echant. Echant. CE. Etchette CE.</p>
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## Packaging

Our standard buffer solutions are bottled in HDPE bottles and delivered in a plastic bag together with its certificate of analysis.



## 1. Traceability

All our standard buffer solutions are directly traceable to standard reference materials from NIST. We buy certified primary standard reference materials from NIST (National Institute of Standards and Technology, USA) and we measure our standard buffer di-rectly against them. This assures correct traceability to NIST.

## 2. Uncertainty

The total uncertainty factor of our standard buffer solutions is max. ± 0,01 pH units, except solutions pH 10, 11, 12 and 13, where two point calibration is performed and uncertainty is max. ± 0,02 pH units.

**Scharlau Chemie**  
 Tel. int.: +34-93-7151811  
 Email: lab@scharlau.com

**CERTIFICATE OF ANALYSIS**

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**Product:** Buffer solution pH = 4,00 (20 °C) (Potassium hydrogen phthalate)  
**SO1004**

Batch 76496  
 Test date 1/2/2006  
 Shelf life 2/2008

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Analysis	Batch value	Guaranteed value
pH at 20 °C uncertainty ± 0,01	4,004	4,00

**Preparation**  
 Standard buffer solutions are prepared using gravimetric and volumetric procedures.  
 Composition per litre is 10,21g Potassium hydrogen phthalate  
 Contains preservative

**Temperature dependence of the pH value**  
 When calibrating your pH-meter at different temperatures than 20°C, refer to the table below to introduce accurate pH values.

T (°C)	pH
0	4,00
5	4,00
10	4,00
15	4,00
20	4,00
25	4,01
30	4,02
35	4,03
40	4,04
45	4,05
50	4,06

**Traceability**  
 This pH buffer solution is traceable to Standard Reference Material from NIST  
 SRM 185h Potassium hydrogen phthalate,  
 SRM 185g Phosphate Buffers,  
 SRM 188 Potassium hydrogen tartrate,  
 SRM 189b Potassium tetroxalate and  
 SRM 2193a Calcium carbonate

**Uncertainty**  
 It characterises the dispersion of the values that could be attributed to the mesurand. The limits of the expanded uncertainty are given at a confidence level of 95% (k=2)

**Measurement**  
 The batch value is determined by measurement with a combination glass electrode against five-point calibration according to DIN 19268. The use of more than five points does not yield any significant improvement in the statistical information obtainable.  
 Calibration standards are prepared according to DIN 19266.  
 Batch value certified at the time of measurement

**Storage and use**  
 For pH-meter calibration  
 If product is stored and unopened, this solution is stable for 2 years from the date of manufacturing.  
 Once the bottle is opened, store tightly closed at room temperature. Avoid exposure to light.  
 Remember to write the date of opening in the space provided in the label for this purpose.  
 Never introduce the electrode in the bottle for measurements.  
 Never pour the used solution back in the bottle.

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This certificate does not release the user from the reception control.

This certificate is an electronic copy of the certificate available in our laboratory, and does not require signature

If you need further details, please call at our factory or contact our local distributor.

*You can get a copy of any of our COA from our web site: [www.scharlau.com](http://www.scharlau.com)*

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page 1 of 1

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### 3. Multi-point calibration

Multi-point calibrations are more precise than two-point or bracketing calibrations. We use five-point calibration whenever possible because the use of more than five points does not yield any significant improvement in the statistical information obtained.

In five-point calibration, the cell electromotive force is determined in five standard buffer solutions and a linear regression calculation is performed.

Measurement is done according to DIN 19268.

### 4. Temperature dependence of the pH

The pH value of a solution depends on the temperature. This is the reason why it is only useful to quote a pH value if the measuring temperature is stated at the same time.

We usually state the pH values of our standard buffer solutions at 20°C, but we also manufacture the most used pH solutions (pH 4, 7 and 10) at 25°C stated temperature.

pH - Temperature dependence tables of our standard buffer solutions are printed in our certificates and labels.

### 5. Date of opening

Our labels contain a specific area for the user to record the date of opening.

## Standard buffer solutions (20°C)

We offer a broad range of solutions from pH 1 to pH 13 (20°C).



Product	Composition	Capacity	Cat. no.
Standard buffer solution pH 1 ± 0,01 (20°C)	Glycine/Sodium chloride/Hydrochloric acid	250 ml	SO11010250
Standard buffer solution pH 1 ± 0,01 (20°C)	Glycine/Sodium chloride/Hydrochloric acid	1 l	SO11011000
Standard buffer solution pH 2 ± 0,01 (20°C)	Citric acid/Sodium hydroxide/Hydrochloric acid	250 ml	SO10220250
Standard buffer solution pH 2 ± 0,01 (20°C)	Citric acid/Sodium hydroxide/Hydrochloric acid	1 l	SO10221000
Standard buffer solution pH 3 ± 0,01 (20°C)	o-Phosphoric acid/Sodium hydroxide	250 ml	SO10230250
Standard buffer solution pH 3 ± 0,01 (20°C)	o-Phosphoric acid/Sodium hydroxide	1 l	SO10231000
Standard buffer solution pH 4 ± 0,01 (20°C)	Potassium hydrogen phthalate	250 ml	SO10040250
Standard buffer solution pH 4 ± 0,01 (20°C)	Potassium hydrogen phthalate	500 ml	SO10040500
Standard buffer solution pH 4 ± 0,01 (20°C)	Potassium hydrogen phthalate	1 l	SO10041000
Standard buffer solution pH 4 ± 0,01 (20°C)	Potassium hydrogen phthalate	5 l	SO1004005P
Standard buffer solution pH 4,01 ± 0,01 (20°C)	Potassium hydrogen phthalate	250 ml	SO10050250
Standard buffer solution pH 4,01 ± 0,01 (20°C)	Potassium hydrogen phthalate	1 l	SO10051000
Standard buffer solution pH 5 ± 0,01 (20°C)	Acetic acid/Potassium hydroxide	250 ml	SO10250250
Standard buffer solution pH 5 ± 0,01 (20°C)	Acetic acid/Potassium hydroxide	1 l	SO10251000
Standard buffer solution pH 6 ± 0,01 (20°C)	Potassium dihydrogen phosphate/Sodium hydroxide	250 ml	SO10060250
Standard buffer solution pH 6 ± 0,01 (20°C)	Potassium dihydrogen phosphate/Sodium hydroxide	1 l	SO10061000
Standard buffer solution pH 7 ± 0,01 (20°C)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	250 ml	SO10070250
Standard buffer solution pH 7 ± 0,01 (20°C)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	500 ml	SO10070500
Standard buffer solution pH 7 ± 0,01 (20°C)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	1 l	SO10071000
Standard buffer solution pH 7 ± 0,01 (20°C)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	5 l	SO1007005P
Standard buffer solution pH 7,02 ± 0,01 (20°C)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	250 ml	SO10080250
Standard buffer solution pH 7,02 ± 0,01 (20°C)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	1 l	SO10081000
Standard buffer solution pH 7,02 ± 0,01 (20°C)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	5 l	SO1008005P
Standard buffer solution pH 8 ± 0,01 (20°C)	di-Sodium tetraborate/Calcium chloride/Hydrochloric acid	250 ml	SO10180250
Standard buffer solution pH 8 ± 0,01 (20°C)	di-Sodium tetraborate/Calcium chloride/Hydrochloric acid	1 l	SO10181000
Standard buffer solution pH 9 ± 0,01 (20°C)	Boric acid/Potassium chloride/Sodium hydroxide	250 ml	SO10090250
Standard buffer solution pH 9 ± 0,01 (20°C)	Boric acid/Potassium chloride/Sodium hydroxide	1 l	SO10091000
Standard buffer solution pH 9 ± 0,01 (20°C)	Boric acid/Potassium chloride/Sodium hydroxide	5 l	SO1009005P
Standard buffer solution pH 10 ± 0,02 (20°C)	Sodium carbonate/Sodium hydrogen carbonate	250 ml	SO10100250
Standard buffer solution pH 10 ± 0,02 (20°C)	Sodium carbonate/Sodium hydrogen carbonate	1 l	SO10101000
Standard buffer solution pH 10 ± 0,02 (20°C)	Sodium carbonate/Sodium hydrogen carbonate	5 l	SO1010005P
Standard buffer solution pH 11 ± 0,02 (20°C)	Boric acid/Sodium hydroxide/Potassium chloride	250 ml	SO11410250
Standard buffer solution pH 11 ± 0,02 (20°C)	Boric acid/Sodium hydroxide/Potassium chloride	1 l	SO11411000
Standard buffer solution pH 12 ± 0,02 (20°C)	di-Sodium hydrogen phosphate/Sodium hydroxide	250 ml	SO11420250
Standard buffer solution pH 12 ± 0,02 (20°C)	di-Sodium hydrogen phosphate/Sodium hydroxide	1 l	SO11421000
Standard buffer solution pH 13 ± 0,02 (20°C)	Potassium chloride/Sodium hydroxide	250 ml	SO11430250
Standard buffer solution pH 13 ± 0,02 (20°C)	Potassium chloride/Sodium hydroxide	1 l	SO11431000

## Coloured standard buffer solutions

The coloured solutions are easily identified by the users and avoid mistakes in the laboratory due to a wrong buffer selection. They are also widely used in field analysis.

We offer coloured solutions measured at 20°C and 25°C.



### Coloured standard buffer solutions (20°C)

Product	Composition	Capacity	Cat. no.
Standard buffer solution pH 4 ± 0,01 (20°C) (red)	Potassium hydrogen phthalate	250 ml	SO20040250
Standard buffer solution pH 4 ± 0,01 (20°C) (red)	Potassium hydrogen phthalate	1 l	SO20041000
Standard buffer solution pH 7 ± 0,01 (20°C) (yellow)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	250 ml	SO20070250
Standard buffer solution pH 7 ± 0,01 (20°C) (yellow)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	1 l	SO20071000
Standard buffer solution pH 10 ± 0,02 (20°C) (blue)	Boric acid/Potassium chloride/Sodium hydroxide	250 ml	SO20100250
Standard buffer solution pH 10 ± 0,02 (20°C) (blue)	Boric acid/Potassium chloride/Sodium hydroxide	1 l	SO20101000

### Coloured standard buffer solutions (25°C)

Product	Composition	Capacity	Cat. no.
Standard buffer solution pH 4 ± 0,01 (25°C) (red)	Potassium hydrogen phthalate	250 ml	SO30040250
Standard buffer solution pH 4 ± 0,01 (25°C) (red)	Potassium hydrogen phthalate	1 l	SO30041000
Standard buffer solution pH 7 ± 0,01 (25°C) (yellow)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	250 ml	SO30070250
Standard buffer solution pH 7 ± 0,01 (25°C) (yellow)	Potassium dihydrogen phosphate/di-Sodium hydrogen phosphate	1 l	SO30071000
Standard buffer solution pH 10 ± 0,02 (25°C) (blue)	Boric acid/Potassium chloride/Sodium hydroxide	250 ml	SO30100250
Standard buffer solution pH 10 ± 0,02 (25°C) (blue)	Boric acid/Potassium chloride/Sodium hydroxide	1 l	SO30101000

*All our standard buffer solutions are delivered together with its certificate of analysis  
Shelf life of our standard buffer solutions is typically 2 years.*

How many times did you dispose of an unfinished buffer bottle because you were not sure of its accuracy?

If your answer is a number, you need MONOBUF.



## NEW MONOBUF

Ready-to-use standard buffer solutions packaged in single doses.

Our MONOBUF packaging allows you to open a new buffer solution "bottle" every time you perform a calibration.

### Without MONOBUF:

1. Open a new bottle of standard buffer solution.
2. Write on the label the date of opening.
3. Pour the solution from the original bottle to a smaller measuring vessel.
4. Label the vessel and write down the pH and date to keep it identified in the laboratory. Often this solution is used for a number of calibrations for a few days.
5. Do the measurement.

### With MONOBUF everything is easier:

1. Take one of the 30 ml containers in the MONOBUF box.
2. Write down the date of opening on the label.
3. Do the measurement.



Description	Capacity	Cat. no.
MONOBUF pH 4 ± 0,01 (20°C) (red)	12 x 30 ml	SO20400360
MONOBUF pH 7 ± 0,01 (20°C) (yellow)	12 x 30 ml	SO20700360
MONOBUF pH 10 ± 0,02 (20°C) (blue)	12 x 30 ml	SO21000360

CAT-BUFEN6

*"Use fresh buffer solutions to calibrate your pH-meter". This is always recommended by pH-meter manufacturers.*